

ABSOLUTE CONFIGURATION OF THE BENZOFURANOID NEOLIGNANS*

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Key Word Index—2,3-Dihydrobenzofuranoid neolignans; 2,3,3a,6-tetrahydrobenzofuranoid neolignans; 2,3,5,6-tetrahydrobenzofuranoid neolignans; absolute configurations.

Abstract—The benzofuranoid neolignans, classified into structurally homogeneous groups by constitution and ORD curves, have had their relative configurations established by ^1H and ^{13}C NMR, including LIS, as well as by X-ray crystallography. Chemical inter-relations of several compounds, which ultimately gave (2*S*,3*S*)-, (2*R*,3*R*)-, and (2*R*,3*S*)-2,3-dihydrobenzofurans, have shown what the absolute configurations are.

INTRODUCTION

The tentative assignment of configuration to C-5 in the 2,3,5,6-tetrahydrobenzofuran neolignans (types 7–9) was based solely on biogenetic reasoning [2]. In the present paper, we propose a chemical solution to this problem and report comprehensively on the absolute stereochemistry of the benzofuranoid neolignans of types 1–9.

RESULTS

Chemical and ORD data

The absolute configurations of the 2*S*, 3*S* (1*a*–*g*) and the 2*R*, 3*R* (2*a*, *b*) groups of *trans*-2,3-dihydrobenzofuran neolignans were established by correlation with representatives of established structure [3] by ORD curves. Acid isomerization of 3*a* (negative Cotton effect at 290 nm) led to 1*d* (positive Cotton effect at 290 nm). Clearly this reaction proceeds by epimerization only of C-2, and defines the absolute configuration of 3*a* [4]. Since all natural and synthetic *cis*-dihydrobenzofuran neolignans (3*a*–*f*) show negative Cotton effects, they must all possess the 2*R*,3*S*-configuration.

For the 2,3,3a,6-tetrahydrobenzofuran neolignans 4*a*–*d*, 5*b*, 6*a*–*c* ^{13}C NMR data [5], confirmed for 5*b* and 6*b* by X-ray crystallography (see below), indicate the relative stereochemistry of the 2,3,3a-bonds. Hydrogenolytic (4*a* → 1*h*) [3], pyrolytic (4*a* → 1*f*, *h* [2]; 5*b* → 1*l* [6]; 6*b* → 3*a*, *c*, *d*; 6*c* → 3*e*, *f* [2]) and photolytic (5*b* → 1*l* [6]) conversions to 2,3-dihydrobenzofurans of established stereochemistry allow, consequently, assignment of

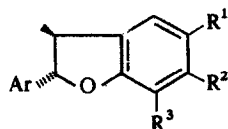
absolute configuration to 4*a*, 5*b*, 6*b* and 6*c*. The Cotton effects related to the aryl (295 nm) and the dienone (260 nm) chromophores of these compounds were analysed and found to be positive in the cases, respectively, of 2*S* (4*a*, 5*b*) and 3*aS* (5*b*, 6*b*, 6*c*) configurations; and negative in the cases, respectively, of 2*R* (6*b*, 6*c*) and 3*aR* (4*a*)-configurations. Proof of the correctness of UV/ORD correlations was seen in the fact that acid isomerization of 5*b* to 6*d* was accompanied solely by inversion of the Cotton effect around 312 nm [6]. The ORD data were used subsequently in the assignment of absolute stereochemistry to the remaining compounds of types 4–6.

The absolute configuration of the heterocycles of compounds 8*c* and 8*d* were again deduced by pyrolytic conversion, respectively to 1*g*, *i* and 1*e*, *j*, *k*, *l*, 2,3-dihydrobenzofurans of established stereochemistry [2]. The negative Cotton effect related to the aryl chromophore (280 nm) of 8*c*, *d*, and of the other compounds of type 7 and 8, thus signals a 2*S*-configuration, and the positive Cotton effect at 280 nm of compounds of type 9 signals a 2*R*-configuration. The stage was thus set for the solution of the problem concerning the chirality at C-5. Indeed, the absolute stereochemistry of 4*a* and 6*b* being known, their thermal Cope rearrangement products, which must originate by suprafacial allyl migration, can only be represented respectively by 7*a* [7] and 9*b* [4]. Thus, again, the positive Cotton effect related to the dienone chromophore (315 nm) of 7*a*, and of the other compound of this type 7*b*, signals a 5*S*-configuration, and the negative Cotton effect at 315 nm of 9*b*, and the other compounds of type 8 and 9, signals a 5*R*-configuration. The relative stereochemistry of 8*d* was confirmed by X-ray crystallography (see below).

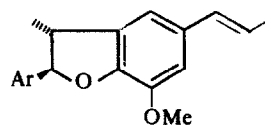
LIS data

The NMR shift reagent $\text{Pr}(\text{fod})_3$ associates strongly with carbonyls. It was hoped, therefore, that the relative geometry of the substituents around the CO-group of the oxo-tetrahydrobenzofuran neolignans (types 4–9) might

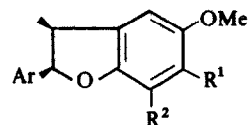
* Part 43 in the series 'The Chemistry of Brazilian Lauraceae'. For Part 42 see Ref. [1]. Sponsored by Fundação de Amparo à Pesquisa do Estado de São Paulo and the CNPq-NSF international exchange program in crystallography. || On leave of absence from Faculdade de Filosofia, Universidade Estadual Paulista, 14800 Araraquara, São Paulo, Brasil.



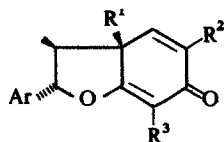
	Ar	R ¹	R ²	R ³	
1a*	Gu	(E)-Pr	H	OMe	[9]
1b*	Pi	(E)-Pr	H	OMe	[9]
1c*	Pi	OMe	OAl	H	[10]
1d*	Mp	OMe	OAl	H	[4]
1e*	Tp	OMe	OAl	H	[2]
1f*	Pi	OMe	OH	Al	[10]
1g*	Mp	OMe	OH	Al	[2]
1h	Pi	OMe	OH	H	[3]
1i	Mp	OMe	OH	H	[2]
1j	Tp	OMe	OH	H	[2]
1k	Tp	OMe	OH	Al	[2]
1l	Tp	Al	OH	H	[2]



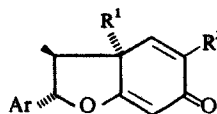
	Ar	
2a*	Ve	[11]
2b*	Tp	[6]



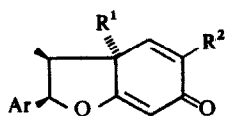
	Ar	R ¹	R ²	
3a*	Mp	OAl	H	[2]
3b*	Pi	OH	Al	[10]
3c*	Mp	OH	Al	[2]
3d	Mp	OH	H	[2]
3e	Tp	OH	H	[2]
3f	Tp	OH	Al	[2]



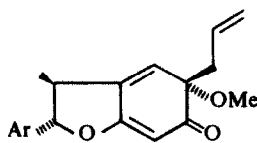
	Ar	R ¹	R ²	R ³	
4a*	Pi	Al	OMe	H	[3]
4b*	Ve	Al	OMe	H	[11]
4c*	Ve	Al	OMe	OMe	[11]
4d*	Tp	OMe	Al	H	[6]



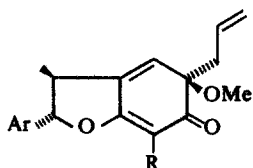
	Ar	R ¹	R ²	
5a*	Tp	Al	OMe	[2]
5b*	Tp	OMe	Al	[6]



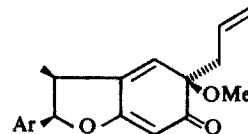
	Ar	R ¹	R ²	
6a*	Pi	Al	OMe	[10]
6b*	Mp	Al	OMe	[2]
6c*	Tp	Al	OMe	[2]
6d*	Tp	OMe	Al	[6]



	Ar	
7a*	Pi	[7]
7b*	Ve	[7]



	Ar	R	
8a*	Pi	H	[10]
8b*	Ve	H	[11]
8c*	Mp	H	[2]
8d*	Tp	H	[2]
8e*	Tp	OMe	[4]



	Ar	
9a*	Pi	†
9b*	Mp	[4]

Gu = Guaiacyl
 Pi = Piperonyl
 Mp = *O*-Methyl-*O*,*O*-methylenylpyrogallyl
 Tp = Tri-*O*-methylpyrogallyl
 Ve = Veratryl
 Pr = Propenyl
 Al = Allyl

* Natural products. † Compound 9a, originally detected only by spectral means as an impurity of 8a [10], was now obtained pure and will be described in full in a forthcoming publication.

Table 1. Wavelength (nm) of ORD extrema and of UV maxima of neolignans of structural types 1–9

	C-2	C-3	C-3a	C-5	ORD			enone obs.	UV band calc.	benzenoid band
					pk	tr	0			
1a, b	S	S			249 ± 1	297 ± 3	274 ± 6			
1c-l	S	S			291 ± 6	316 ± 8	302 ± 5			
2a, b	R	R			301 ± 6	250 ± 5	280 ± 0			
3a-f	R	S			314 ± 7	282 ± 6	295 ± 10			
4a-c	S	S	R		316 ± 13	288 ± 8	303 ± 10	264 ± 7	262	294 ± 9
4d	S	S	R		335	290	312	280		300
5a	S	S	S		282	320	301	257	262	285
5b	S	S	S		335	275	312	280		300
6a-c	R	S	S		284 ± 9	319 ± 2	302 ± 3	258 ± 2	262	293 ± 8
6d	R	S	S		291	335	312	280		300
7a, b	S	S		S	342 ± 1	290 ± 2	312 ± 0	315 ± 0	320	260 ± 10
8a-e	S	S		R	281 ± 11	347 ± 6	317 ± 5	315 ± 2	320	283 ± 5
9a, b	R	S		R	295 ± 0	344 ± 1	316 ± 1	316 ± 1	320	290 ± 0

be amenable to confirmation by a LIS study. Indeed, relatively strong Δ values were observed for the OCH_3 -5 or CH_2 -5 signals of compounds of types 4–6 in which OMe or CH_2 and CO are coplanar (Table 2). The carbonyl is probably tilted towards the α -face of the molecules, since the Δ values for $\text{CH}_2\text{CH}=\text{CH}_2$ and H-2 signals are larger in compounds 6a–c, which have these units on this face, than in compound 4a, which sustains these units on the β -face. This is clearly also the case for compounds of type 7–9 [cf. in Table 2 $\Delta\text{H-2}$ for 9a (Ar- β) vs. 7a, 8a, c, d (Ar- α) and $\Delta\text{H-6'}$ for 8a (Ar- α) vs. 9a (Ar- β)]. The values of OCH_3 -5 and CH_2 -5, however, are of the same order of magnitude and the relative configuration at C-5 is not amenable to analysis by the LIS method.

X-ray crystallographic data

Constitutions and relative stereochemistries of 5b, 6b and 8d were confirmed by direct X-ray crystallographic analyses. Crystals of 5b are monoclinic, space group $P2_1$, with $a = 10.635$ (3), $b = 10.021$ (3), $c = 10.708$ (3) Å, $\beta = 115.72$ (2)° and $Z = 2$. Crystals of 6b are ortho-

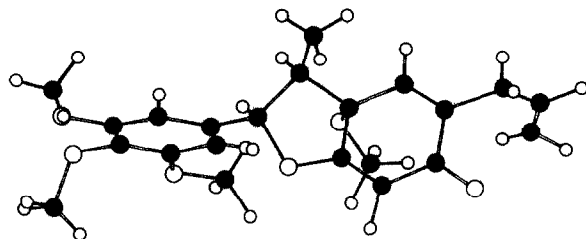
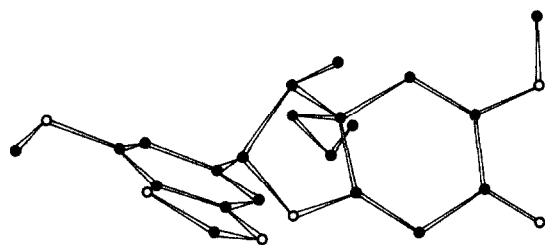
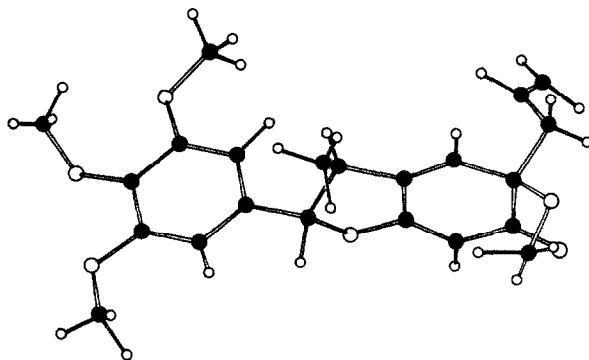
rhombic, space group $P2_12_12_1$, with $a = 7.77$, $b = 11.36$, $c = 21.47$ Å and $Z = 4$. Crystals of 8d are orthorhombic, space group $P2_12_12_1$, with $a = 8.532$ (4), $b = 9.588$ (4), $c = 25.260$ (19) Å and $Z = 4$. The intensities of 2222 (5b), 1139 (6b) and 2432 (8d) independent reflections were measured with an automatic four-circle diffractometer using graphite monochromated Cu K α radiation. The phase problem was solved by direct methods. The E-map of the best set gave the positions of 23 (5b), 19 (6b) and 25 (8d) atoms. A difference Fourier was used to locate 5 (5b), 8 (6b) and 3 (8d) additional atoms. Several cycles of full matrix least square refinement considering anisotropic temperature factors reduced the R factors to 0.079 (5b) and 0.81 (8d). For 5b and 8d, difference Fourier calculations gave positions of some of the hydrogens and a Generation of Atomic Positions program [8] was used to locate all remaining hydrogen atom positions. Refinement of all parameters using isotropic temperature factors for hydrogen gave final R factors of 0.044 (5b) and 0.040 (8d). For 6b, the approximate locations of the 24 hydrogen atoms were determined by orbital geometric considerations. Several cycles of full matrix anisotropic

Table 2. LIS Δ values (ppm) recorded for proton signals of oxo-tetrahydrobenzofuranoid neolignans*

Compound	Benzofuran		CH ₂ CH= system				Aromatic				ring			
	OMe	H-7	H-4	CH ₂	H-2	H-3	Me-3	H-2'	H-5'	H-6'	O ₂ CH ₂ -3',4'	OMe-3'	OMe-4'	OMe-5'
4a	27.0	11.7	9.1	4.6	2.4	3.3	2.7	0.7	0.1	0.8	0.2	—	—	—
5b	2.3	8.5	2.8	8.0		2.4	2.0	2.7	—	2.7	—	2.7	5.0	2.7
6a	28.2	14.2	10.4	5.6	3.4	4.4	4.0	2.0	0.8	2.0	0.4	—	—	—
6b	28.4	14.3	11.7		3.5		4.0	2.0	—	2.0	0.4	—	—	0.4
6c	30.0	13.4	10.6		3.5		3.5	1.8	—	1.8	—	0.6	0.8	0.6
7a	10.7	8.9	5.8	10.2	2.2	2.8	1.6	0.6	0.3	0.7	0.1	—	—	—
8a	11.8	11.2	7.0	10.4	1.6	2.8	2.2	0.8	0.8	1.6	0.6	—	—	—
8c	12.0	9.4	6.8	8.4	1.8	3.0	1.8	1.0	—	1.5	0.4	—	—	—
8d	8.0	8.6	5.4		1.8	4.6†	2.0	2.4	—	2.4	—	3.0	5.6	3.0
9a	11.7	9.2	8.7	9.8	3.0	3.0	2.1	0.8	0.8	0.8	0.4	—	—	—

* Shift studies were carried out by stepwise addition of known amounts of Pr(fod)₃ to ca 0.15 M solutions of substrate in CDCl₃. The LIS data were obtained by graphic extrapolation of observed shifts to 1:1 molar shift reagent–substrate ratio.

† Association of reagent with CO and trimethoxyaryl is responsible for, respectively, 2/3 and 1/3 of this value.

Fig. 1. X-Ray structure of **5b**.Fig. 2. X-Ray structure of **6b**.Fig. 3. X-Ray structure of **8d**.

least-squares refinement, with fixed isotropic thermal parameters for the H atoms, reduced the R factor to 0.098. Details of the X-ray structural analyses will be published elsewhere.

The molecular structures found in the crystals of **5b**, **6b** and **8d** are shown respectively in Figures 1, 2 and 3.

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